

Naming Ionic Compounds

Write the NAME of the compounds below using correct nomenclature rules.

1. Na_2O _____

2. NaCl _____

3. MgBr_2 _____

4. KCl _____

5. Al_2S_3 _____

6. Li_3N _____

7. Mg_3P_2 _____

8. K_2O _____

9. MgCl_2 _____

10. Ca_3P_2 _____

11. Be_2SO_4 _____

12. NaOH _____

13. $\text{B}_3(\text{OH})_2$ _____

14. KNO_3 _____

15. MgSO_4 _____

16. KMnO_4 _____

17. AlPO_4 _____

18. LiOH _____

19. CaSO_4 _____

20. NH_4I _____

21. FeCl_2 _____

22. CrF_2 _____

23. PbO _____

24. Co_3N_2 _____

25. FeCl_3 _____

26. Cu_2S _____

27. $\text{Cu}(\text{NO}_3)_2$ _____

28. FePO_4 _____

29. FeSO_4 _____

30. CoCO_3 _____